

MEMBRANE FILTRATION TECHNOLOGY FOR WATER TREATMENT IN TERMS OF MEMBRANE PERFORMANCE, ENERGY CONSUMPTION, AND COST ASPECTS: A REVIEW

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ABSTRACT

The problems of water quality and its management are a global issue that must be resolved. The membrane filtration technology shows good potential for water treatment to improve water quality, eliminate harmful contaminants, and efficiently restore nutrients. This article aims to comprehensively review membrane filtration, including membrane filtration performance to remove contaminants, energy consumption, cost aspects, and antifouling strategies. Technologies in membrane filtration are microfiltration (MF), ultrafiltration (UF), nanofiltration (NF), and reverse osmosis (RO). This review systematically evaluates recent advancements in membrane filtration technologies, particularly focusing on membrane performance, energy consumption, and water treatment costs. The data for this review were collected from a comprehensive range of sources, including peer-reviewed journal articles, conference proceedings, and industry reports. The results of this study indicate that the MF/UF system can effectively remove bacteria and be used as a pre-treatment before the NF/RO process. NF/RO system can remove contaminants higher than MF/UF. It can remove almost all contaminants, ions, heavy metals, viruses, and pharmaceutical compounds (PhAC). However, the system is more susceptible to membrane fouling, but adequate pre-treatment can solve these problems. Water treatment with NF and RO offers a good alternative because its operation makes it possible to use lower energy consumption than conventional water treatment. Based on the cost aspect, if the virus and ion contamination are not the main target in the filtering and not for consumption, the MF/UF system is sufficient. It will be more economical than using NF/RO. The findings of this study can provide insight into the effectiveness of membrane filtration technology in water treatment. Future research can develop membrane filtration technology for good performance, low energy consumption, and low cost.

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1. INTRODUCTION

In the past decade, many people have made awareness and progress in water treatment to get high-quality drinking water. However, approximately 663 million people do not get high-

quality drinking water. They consume untreated water from wells, springs, and surface water (UNICEF, 2015). The World Economic Forum 2019 also said freshwater scarcity is one of the biggest global problems (Forum, 2019). Thus, water treatment is needed to treat surface water into fresh water that can be consumed. However, if the surface water is not treated or repaired, it can infect consumers with various diseases.

Another problem is poor sanitation, which causes pathogens to contaminate surface and groundwater. Contaminated drinking water can cause consumers to become infected with diseases like diarrhea and vomiting (L. Lin et al., 2022; Shah et al., 2023). Water-borne diseases can be prevented by proper water treatment. It is important to apply low-cost water filtration techniques to treat contaminated water. Generally, conventional activated sludge processes treat water biologically by removing organic substances/ nutrients. According to WHO's water standards, the microfiltration (MF)/ ultrafiltration (UF) process continues to produce high-quality permeated water. The treated water is ready to drink. The next process is a high-pressure membrane process, such as nanofiltration (NF) or reverse osmosis (RO), to remove organic and dissolved ions (Changmai et al., 2020). In a recent study, the NF process was combined with biological processes, and then the NF permeate was processed by RO to obtain water with very high levels of purification (Tay et al., 2018).

However, biologically, water treatment processes are unsuitable for all types of contaminated water due to low treatment efficiency. In addition, biological-based processing is highly dependent on environmental factors such as temperature, variations in feed composition, oxygen levels, etc. (Meena et al., 2019). Biological aerobic-based water treatment requires much energy for aeration. Some organic/ nutrients in contaminated water are converted to CO₂, leading to large greenhouse gas emissions (Huang et al. 2017). Some other organic/ nutrients in contaminated water are converted to biomass, which is generally used for biogas production (energy recovery) through anaerobic digestion. However, the efficiency of converting organic/ nutrition into biomass is lower than converting organic/ nutrient into CO₂ (Lateef et al. 2013). In addition, the removal of nutrients such as nitrogen (N) and phosphorus (P) during treatment of contaminated water is lower than its recovery.

Research on water treatment using membrane filtration in several types of driving forces (pressure-driven and osmosis-driven) received great attention, especially on improving membrane performance. In a direct membrane filtration system, the membrane effectively rejects organics/ nutrients from contaminated water, so superior permeate quality with a high water recovery ratio was reached. Direct membrane filtration systems generally have a high compactness. The direct membrane process does not involve an additional activated sludge process, which reduces energy consumption. In addition, organic and rejected nutrients enriched at high concentrations can be digested afterward to produce renewable energy (H₂, CH₄) or become fertilizer (Kimura et al. 2017, Xiong et al. 2019). Thus, the direct membrane process can facilitate water treatment properly.

RO has been used among developed countries such as Saudi Arabia and Israel (Harrington, 2013). This process was implemented in the 1990s to replace distillation due to high energy consumption (Cohen-Tanugi & Grossman 2012). In RO, water is given a high pressure from one side of the membrane. Hydrated sodium and chloride ions will be filtered because the membrane is a very small structural gap through which only water can pass (Economist.com, 2013). In addition, this method can produce an average of 250 million gallons of clean water per day. Several

membrane technology applications have proven less efficient in water purification. For example, low salt rejection is a serious problem in carbon nanotube-based membranes, and high-density CNT structures are difficult to produce (Fornasiero et al., 2010). Another problem in water treatment using membrane technology is membrane fouling, especially with high pressure such as NF / RO (Mi et al., 2020). Although there are many studies on membrane technology for water treatment, there is a lack of comprehensive reviews that simultaneously discuss membrane performance, energy consumption, and operational costs in one integrated study.

The problem that needs to be solved in treating contaminated water is finding an efficient process with high performance in removing contaminants, reducing economic costs, and reducing energy consumption for water filtration and antifouling. For this reason, it is necessary to have an in-depth and detailed discussion regarding this matter. To the author's knowledge, there has not been a comprehensive review discussing the issue. This article aims to comprehensively review the filtration membrane that has recently been researched for water treatment. It comprises membrane filtration performance, energy consumption, cost, and antifouling strategies.

2. METHOD

This review systematically evaluates recent advancements in membrane filtration technologies, particularly focusing on membrane performance, energy consumption, and water treatment costs. The data for this review were collected from a comprehensive range of sources, including peer-reviewed journal articles, conference proceedings, and industry reports. Relevant studies were sourced primarily from established academic databases such as ScienceDirect, Wiley Online Library, and Google Scholar. The search terms "membrane filtration," "energy consumption," "cost analysis of membrane filtration systems," and specific filtration technologies like microfiltration (MF), ultrafiltration (UF), nanofiltration (NF), and reverse osmosis (RO) were employed. These resources were carefully selected to ensure a robust and comprehensive analysis of each technology's effectiveness in various water treatment applications.

The performance of each filtration system was assessed by reviewing studies that report the removal efficiency of contaminants, including bacteria, viruses, heavy metals, ions, and pharmaceutical compounds. This performance evaluation was based on key metrics such as permeate flux, rejection rates, and the extent of membrane fouling observed under various operational conditions. For instance, MF and UF membranes were evaluated for their ability to remove larger particles and bacteria. At the same time, NF and RO systems were analyzed for their effectiveness in removing smaller contaminants like dissolved ions and organic compounds. The review also considered the role of pretreatment methods, including integrating low-pressure systems like MF and UF before applying high-pressure systems like NF and RO, as these can significantly impact overall treatment efficiency.

Energy consumption is a critical factor in determining the feasibility and sustainability of membrane filtration systems. This review compared the energy demands of different membrane technologies by analyzing studies that reported power consumption per cubic meter of treated water. Specifically, high-pressure systems like NF and RO, which require significant energy input to achieve the necessary transmembrane pressures, were compared to the more energy-efficient MF and UF systems. The analysis highlighted how membrane material types, operational parameters, and system configurations influence energy consumption. In particular, the review

focused on advancements in membrane material design and the potential for energy savings through system efficiency improvements, such as using energy recovery devices and alternative driving forces.

Cost analysis was another central aspect of this review, which examined the capital, operational, and maintenance costs of the four membrane filtration technologies. A life-cycle cost approach was applied to understand the long-term economic viability of each technology. The analysis provided insights into the financial challenges associated with large-scale implementation by reviewing case studies and real-world applications. For example, while MF/UF systems tend to have lower initial capital costs and are more economical in treating non-potable water, NF and RO systems offer superior contaminant removal capabilities despite their higher operational and maintenance costs. Additionally, the review addressed the economic implications of membrane fouling, a significant issue in membrane-based systems. Antifouling strategies, including surface modifications, cleaning protocols, and pretreatment processes, were analyzed to determine their effect on extending membrane life, reducing cleaning costs, and improving overall system performance.

3. RESULTS AND DISCUSSION

3.1. Water Sources

Almost all water sources on Earth come from the ocean in the form of salt water; the other is fresh water found in the soil and ice in Antarctica. Most water supplies on Earth come from the oceans (Stephens et al., 2020). However, not all humans can use seawater directly because water from the ocean has high salt levels, so it must be treated first to separate the salt. The other surface water is fresh water that can be utilized more easily than seawater for daily life. There is a process of seawater intrusion. Seawater entry into freshwater aquifer areas causes a decrease in the quality of fresh water, especially for drinking and clean water needs. Excessive intake of groundwater by a pump causes a decrease in the groundwater interface, which causes seawater intrusion. The increase in seawater also pushes seawater into the inland freshwater aquifer. In addition to the surface water described above, groundwater is used for daily life. Groundwater moves in the soil, which is contained in the spaces between the grains of soil and rock cracks.

It is called water gap or fissure water. Water that fills the pore layers of the Earth under the water table is usually called groundwater. Surface water quality is influenced by groundwater and water that flows above the surface due to the full capacity of soil infiltration (runoff). During the rainy season, the phenomenon is reversed, where surface water affects groundwater quality (McLachlan et al., 2017).

The interaction between groundwater and surface water has important implications for the quantity and quality of water and ecological health. The presence of geological, hydrological, and biogeochemical heterogeneity in groundwater and surface water is difficult to characterize through direct observation. It has caused many problems, one of which is flooding. For example, high urbanization can increase surface runoff and prevent groundwater filling during the rainy season. Agricultural activities can also reduce the storage of rainwater on the ground. These phenomena can affect surface water in complex ways that depend on many aspects, including plant type, soil type, and climate (Han et al. 2017, Hocking & Kelly 2016). For example, more natural organic

content is expected in humic waters than non-humic waters. Therefore, the permissible variation of dissolved organic carbon (DOC) concentration depends on the type of water (DEFRA, 2014).

3.2. Water Problems

Water source treatment and clean water supply are the main challenges of the 21st century (Zheng et al., 2023). Contaminants in water can pose significant risks to the environment and human health, so the efficient disposal of these contaminants is very important (Cai et al. 2018, Trinh et al. 2018, Wu et al. 2018). The problem in water is eutrophication. Eutrophication is excess nutrients in water or water pollution by excessive nutrition. Water pollution is contaminated or affected water by organic or inorganic substances (Zhou et al. 2020). This problem happens around the world, both in freshwater and seawater ecosystems. Eutrophication is caused by many things, such as human behavior that is not friendly to the environment and nutrient emissions by companies or industries thrown into the water (Jarvie et al., 2018). Nutrient wastes in water can be from industry (Rachbauer et al., 2024), agricultural fertilizers (Chen et al. 2019, Huang et al. 2017), and husbandry waste (Barcellos et al., 2019). Water that contains many nutrients, such as phosphorus and nitrogen, can trigger the growth of phytoplankton and algae. It also increases aquatic productivity (Lin et al. 2020). If there is a large increase, algae will bloom, which can harm the life of organisms in the waters. Eutrophication will threaten the water ecosystem more in the dry and rainy seasons (Zhang et al. 2017). Increasing temperature, changes in rainfall patterns, and changing hydrodynamics in the dry season contribute to eutrophication's acceleration (Lu et al., 2019).

Another problem caused by eutrophication is the appearance of cyanobacteria in water. These microorganisms can produce and release oxygen (Demoulin et al., 2019) and also produce toxins that are harmful to many high-level organisms such as fish, cattle, and humans (Conklin et al. 2020, D'Agostino et al. 2019, Wisniewska et al. 2019). Thus, it is important to treat wastewater that is contaminated with many nutrients and disinfect water to deactivate bacteria (Rajendran et al., 2018)c. The main water sources in households, agriculture, and industry are lakes and reservoirs. These are susceptible to environmental changes because they have low flow velocity or stagnation. Eutrophication can cause significant problems. Nutrition and Natural Organic Matter (NOM) can be removed by processes such as activated sludge (Rajendran et al., 2018) and sand filtration (Kulkarni et al. 2019, Xu et al. 2019). At the same time, removing microorganisms produced in the water because of the many nutrients can use membrane technology (Chang et al. 2019, Hylling et al. 2019; de Vries et al. 2020).

Another problem that can affect water quality is the presence of heavy metals in the water. Heavy metals in water can produce short-term contamination derived from anthropogenic activities (Sarah et al., 2019). These activities include coal burning, mining, pesticide use, battery production, welding, etc. Over time, ecosystems usually adapt to cope with increasing levels of these heavy metals (Paul, 2017). When organisms absorb heavy metals and metalloids, especially Cadmium (Cd), Zinc (Zn), Lead (Pb), and Arsenic (As), they are not easily excreted (Y. Xu et al., 2019). Therefore, they ultimately bioaccumulate in the food chain (Hao et al., 2019; Sarah et al., 2019). For example, pollution caused by heavy metals in mining is a widespread problem worldwide (Qiao et al., 2020). However, pollution also appears around sediments that are not exploited, such as swamps. These areas are contaminated with many pollutants, such as Cu, Zn,

and As, with concentrations exceeding the provisions of the WHO. High concentrations of heavy metals can occur in surface water, sediment, and surface soil.

3.3. Water Treatment

WHO issued a recommendation regarding contaminant parameters that must be followed so that the water sources are safe for consumption. Water treatment aims to produce water whose quality depends on the treatment process (J. A. Silva, 2023). In developed countries, water is treated in wastewater treatment plants (WWTP) and drinking water treatment plants (DWTP), which implement various pre-treatment steps, including coarse and fine filtration, sedimentation, coagulation/circulation, various filtration methods, activated sludge, primary disinfectants, and residues, etc. (Yang et al., 2017).

Each method usually removes certain groups of pollutants. For example, the activated sludge removes NOM, NO₃, and PO₄ from wastewater (Keene et al., 2017). Coagulation and fluctuation remove ionic/ colloidal materials such as clay particles and dissolved metals (Zaharia et al., 2024). The oxidation of organic matter contributes to the release of carbon dioxide into the atmosphere, which endangers the environment. Heavy metal waste can be removed from wastewater using granular activated carbon (GAC) (Bilardi et al. 2018, Eeshwarasinghe, et al. 2019; Mohammad-pajooch et al. 2018). However, high process optimization is required to achieve optimal cleaning (Sountharajah et al., 2015). RO can be used as an alternative to metal removal. The high pressure used in RO causes higher power consumption (Vital et al., 2018). UF and NF systems can also remove heavy metals and metalloids (Lam et al., 2018). MF, UF, NF, and RO filtration systems can be used to remove contaminants such as bacteria, viruses, and other dissolved metals (Krystynik & Tito 2017, Moussa et al. 2017, Sillanpää et al. 2018).

3.4. Membrane Filtration Methods

Contaminants in water cannot be removed effectively with conventional treatments. They can pose significant risks to both the environment and human health. Contaminants have to be removed effectively. Membrane filtration is a good technique for removing many contaminants in water (Abdel-Shafy & Abdel-Shafy, 2017; Veréb et al., 2019). This technology makes it possible to improve the purification of contaminated water and produce a higher quality permeate than conventional treatment. The advantages of membrane technology as a new approach in wastewater treatment are modular design, automated methods, pore size flexibility, great resistance to many types of contaminants, and good permeate quality (Ezugbe & Rathilal, 2020).

A filtering membrane is a separation technique that uses a membrane in several phases. A filtration membrane consists of a single layer of material, a membrane that allows water to pass through. Technologies including membrane filtration are MF, UF, NF, and RO, which filter 0.5 - 5 µm particles for MF, 0.005 - 0.5 µm particles for UF, and 0.0007 - 0.005 µm particles for NF. RO can eliminate almost all contaminants in water (Global 2012). RO and NF are very effective for removing dissolved ions and organic solutes. However, operating RO and NF membranes requires high pressure (6-70 bar). In contrast, UF and MF membranes require lower pressures (0.3 - 5 psi) than others. UF and MF are ineffective in maintaining dissolved ions and organic solutes (Diallo, 2014).

Membranes are usually made of woven fibers (Phiri et al., 2019), ceramics (Mestre et al., 2019; Myat et al., 2018; Rasouli et al., 2019), and polymeric materials (Norouzi et al., 2019; Veréb et al., 2019; Zaouk et al., 2020). Ceramic membranes have advantages in terms of thermal, chemical, and mechanical stability and longer usage times (Tai et al., 2019). There are several types of polymeric membranes, including polyethersulfone (PES), polyacrylonitrile (PAN), and polytetrafluoroethylene (PTFE). PES has excellent mechanical and chemical resistance, high stability, easy manufacturing, and good environmental durability (Yin & Zhou 2015). PAN has good hydrophilicity, membrane surface roughness, and a strong antibacterial effect (Li et al. 2019). PTFE has exceptional chemical resistance and high porosity (Wei et al., 2017).

They can also be modified to improve performance before being applied in water purification. They can be used to reduce impurities (Nidhi et al. 2019), purify oil emulsions (Veréb et al., 2019), and improve the removal of specific pollutants, such as arsenic (Chatterjee & De 2017), Mg and Ca (Zaouk et al., 2020), bacteria (Li et al. 2019). There is no general agreement on the most beneficial polymer material for filtering contaminants in water. Membrane technology can follow environmental standards and can ensure the reuse of wastewater. Thus, a thorough discussion of water treatment with membrane technology is very interesting to investigate so that the characteristics of each membrane can be known.

3.5. Low Pressure Membrane Filtration (MF dan UF)

MF cannot remove small particles, unlike UF and NF. However, MF effectively removes bacteria (Hellenga et al., 2019; Thuy & Boontawan, 2017). MF can be used in domestic water recycling systems (Choo-bar et al., 2019; Manouchehri & Kargari, 2017). In addition, many MF technology activities are wastewater treatment from the textile industry (Manni et al., 2020) and soy sauce improvement process (Guo et al., 2020). Low-pressure membranes (MF and UF) can also be applied as a pretreatment system for high-pressure membranes (NF and RO) (Changmai et al., 2020; Tawalbeh et al., 2018; X. Zhang et al., 2020). The aim is to reduce the possibility of ineffectiveness in the NF/RO process because it can eliminate the potential of pollutants such as bacteria (Hellenga et al., 2019) and improve the stability of NF/RO performance (Al-Mashharawi et al., 2013). In the pretreatment cycle, MF can remove bacterial cells, while other dissolved impurities, such as proteins, colloids, dyes, and salts, will be removed in the NF process (Thuy & Boontawan 2017).

The materials used in MF/UF membrane technology are ceramics, polymers, and woven fabrics. These materials can be modified with other materials, such as titanium dioxide (TiO₂) nanoparticles or polyvinylpyrrolidone, to improve hydrophilicity and disinfection (Carpintero-tepole et al. 2017). The MF process can also be integrated with pre-coagulation (Almojjly et al., 2019; Hao et al., 2019; Hassan et al., 2019). This process can remove chemical compounds, waste oil, and ARGs (Antibiotic resistance genes) from the water. ARGs have been considered pollutants that appear in WWTP wastes. It is because of the potential risks to human health and ecological safety when reused. The results show that the integrated process can reduce the absolute abundance of total ARGs and eARGs from waste. The integrated process performs well in removing common pollutants (for example, organic carbon and dissolved phosphate) from waste to improve water quality. The integrated process reduces membrane impurities more than MF as a single process. Pre-coagulation and MF integration is an advanced wastewater treatment technology that can

remove ARG (especially eARG) from WWTP waste for reuse. Another integration is electrocoagulation (EC), which is followed by the MF process (Changmai et al., 2020). This is effectively used to remove metal (Mn and Fe) contaminants.

Other recent research is the manufacture of filtration membranes from natural magnesite (Manni et al., 2020). Magnesite added with clay is optimized to be a ceramic membrane as an MF membrane for filtering wastewater in the textile industry. The integration is intended to increase mechanical strength and reduce sintering temperatures. The experimental results show that the magnesite membrane effectively cleans textile wastewater. The filtration results show that the membrane efficiently removes all turbidity from textile wastewater (99.9%) while eliminating chemical oxygen demand (COD) by 69.7%.

UF is a selective separation process that purifies water, such as proteins, carbohydrates, and enzymes (Global 2012). UF is also very effective in removing microorganisms in wastewater, especially protozoa (*Cryptosporidium*) and bacteria (*Escherichia coli*) (Ferrer et al., 2015). However, the pore size is not small enough to eliminate norovirus (Yasui et al., 2016), adenovirus (Z. Yin et al., 2015), rotavirus (Qiu et al., 2015), and bacteriophages (Ferrer et al., 2015). Virus removal can be improved by using coagulation as a pretreatment for UF (Lee et al. 2017). The efficiency of coagulation can be dramatically improved by optimizing the pH. Coagulation as a pretreatment in UF can also reduce fouling drastically (Racar et al., 2019).

Regarding fouling, UF and MF are vulnerable to fouling even though it is less severe than NF or RO. Recent research shows that Novel Polyetherimide composite membranes in UF can effectively antifouling (Sathya et al., 2020). Antifouling in the study was observed using model pollutants such as bovine serum albumin, humic acid, and real textile wastewater. Polysulfone-based UF membranes with dopamine and Nisin are also very effective in overcoming fouling (Kim et al., 2018). Nisin, a low molecular weight antimicrobial peptide, is immobilized on the surface of the UF membrane. The resulting Nisin-containing UF membrane shows good fouling resistance and flux recovery ability.

3.6. High Pressure Membrane Filtration (NF dan RO)

NF can be used to remove solutes greater than RO. The size of the contaminant removed is in the range between UF and RO. NF effectively removes particles between sizes of 100 and 1000 Da (Z. Li et al., 2013). Despite these advantages and the application in this case, drinking water production, wastewater treatment, the food industry, the chemical and pharmaceutical industries, and many other industries, there are an inhibit the application of NF and RO on a large scale, namely fouling (Mi et al., 2020; F. C. Silva, 2018; X. Zhu et al., 2020). Fouling can reduce membrane performance due to particle accumulation and microbial growth on the membrane surface (Peña et al., 2013). RO membranes are more susceptible to fouling than NF membranes (Foureaux et al., 2019). Fouling can be reduced by adequate pre-treatment. This pre-treatment is in the form of conventional pre-treatment (coagulation, flocculation, sedimentation) or pre-treatment MF and UF. However, pre-treatment cannot eliminate all microorganisms that pollute the membrane (Badruzzaman et al., 2019). Most membrane installations must be cleaned to remove recalcitrant bio-films and maintain product quality (Beyer et al., 2017).

Many recent studies discuss antifouling in high-pressure filtration membranes. Nanocomposite membranes with natural clay bentonites are used in the polyamide selective layer

as flux enhancers and antifouling agents in NF and RO systems (Nidhi Maalige et al., 2019). Sulfonated clays are induced in the polyamide selective layer through a modified interface polymerization method. Incorporating these nanoparticles results in the formation of membranes with enhanced flux, rejection of many contaminants, and the induction of anti-impurities in conventional polyamide layers. Other studies on developing antifouling NF membranes include zwitterionic functionalized monomers (Mi et al., 2020). The zwitterionic membrane structure can be adjusted for separation performance and antifouling properties. The developed membrane shows antifouling performance with a flux recovery ratio of 94.9%. Antifouling can be solved by infusing goethite-tannic acid nanoparticles (Saniei et al., 2020). Nanoparticles are synthesized through the phase inversion method. Membranes modified by 0.5% by weight of nanoparticles showed the best performance in overcoming fouling.

NF and RO can remove most solutes and produce stable and clean water. NF and RO have become interesting things in water treatment. Improved material design has led to advances in membrane functionality (permeability and selectivity) and membrane applicability (mechanical and chemical stability) (Salman et al., 2022). In a single purification step, NF and RO provide options to eliminate most impurities, such as hardness, color, and disinfection products (Werber et al., 2016). Other advantages include low purchase costs and space requirements, and NF/RO systems can be easily upgraded and operated continuously and automatically (S. Lin & Elimelech, 2015). This explains that NF and RO membrane filtration has become the most important technology for seawater desalination.

NF is less effective in filtering ions from water than RO, but operating and maintenance costs can be lower (Golpour & Pakizeh, 2017; Moraes et al., 2018). This is because it has lower TMPs (Transmembrane Pressure), producing water flows with the same permeate. For certain contaminants, such as pharmaceutical compounds (PhAC), NF generally shows better eliminating ability than UF but less than RO (Foureaux et al., 2019; Licona et al., 2018). UF, NF, and RO can be chemically adjusted to enhance the removal of specific contaminants. For example, NF can remove pharmaceutical compounds with a pressure of 20 bar and pH 5. The situation can reject the nonionic compound acetaminophen and caffeine more than 90%. As for the anionic compound of the mother-proven, dipiron and diclofenac, using a pH of 5 or 7 (Licona et al., 2018).

Based on membrane performance, the NF/RO system can remove almost all contaminants (A. S. C. Chen et al., 2020; Gholami et al., 2020; Peydayesh et al., 2019). Hybrid NF membrane with Polyethyleneimine is the recent research in water treatment. Hybrid membranes have superior performance in removing Zn (99.06%), Mg (97.36%), Cd (96.72%), Cu (95.84%), Ca (95.25%), Ca (95.25%), Ni (94.63%), and Pb (93.39%) (Peydayesh et al., 2019). Hybrid membranes also show good antifouling properties. Membrane fouling can be overcome by reducing the surface roughness of the membrane, as well as surface energy. Most fouling adsorption occurs on the valley surface membrane. For certain parameters, for example, pH = 2, pressure = 10 bar, and permeate recovery rate = 45%, can remove 94% sulfate contaminants, 98% calcium, 85% As III, and 75% As V with reusable permeates accordingly WHO standards (Reis et al., 2019). Using sulfonated poly-equipped NF as a new interlayer polymer can reject 99.7% Na₂SO₄ at pressure = 10 bar (Y. Zhu et al., 2020).

Water treatment with NF and RO offers an attractive alternative because its operation uses lower energy consumption than conventional treatment. This has had an impact in the last decade, when NF and RO took over the market share in water treatment. RO is a membrane separation

process that offers more energy savings than conventional separation processes (Dong et al., 2020). Energy efficiency can be a critical consideration for industrialization due to significant long-term savings in energy costs (S. Lin & Elimelech, 2015). The minimum energy requirement in the conventional distillation process is 225 kJ/mol; pervaporation requires 75 kJ/mol. In contrast, the minimum energy requirement for RO with membrane organosilica is lower than that for conventional distillation or pervaporation, which is 2.3 kJ/mol (Dong et al., 2020). These results indicate that RO has great potential to separate organic solvent mixtures and is efficient in energy consumption. Membranes have excellent selectivity, superior solvent resistance, and adequate mechanical strength.

There is a recent study in which solar energy operates the NF/RO system to treat brackish water, so it is suitable for consumption (Richards et al., 2015; Shen et al., 2016, 2019). The main performance parameters investigated are flux, fluoride retention, and specific energy consumption (SEC). The research compares 1×4 inch module, one element in 4 inches NF/RO (BW30), to a 3 × 2.5-inch module, three elements in 2.5-inch NF/RO (BW30, BW30LE, and NF90) (Shen et al., 2019). The results show that the 1 × 4-inch module's overall performance is superior to the 3 × 2.5-inch module. The 3 × 2.5-inch module performance decreases significantly from the first element to the third element. It is because of an increase in feed concentration and a decrease in net driving pressure. All of them can follow WHO drinking water recommendations related to fluoride content. The NF90 membrane showed good performance, balanced productivity, and permeate quality. There are 1582 L of clean drinking water produced with specific energy consumption averaging 1.6 kWh/m³ for one day (Shen et al., 2016). Solar-powered membrane systems equipped with modules provide a good solution to overcoming the drinking water problem in rural areas.

3.7. Cost Aspects in Water Treatment

The operational and maintenance costs of MF, UF, NF, and RO technologies vary greatly. They depend on the variability of several factors, including plant capacity, assumptions, and scale at the time of the study. However, in general, they can be compared in terms of capital, operational, and maintenance costs.

Suppose viruses and ion contamination are not the main targets. In that case, the filtering system can only use MF/UF by providing acceptable quality efficiencies and is more economical than using a combination of MF/UF and NF/RO. For example, *Chlorella minutissima* can be removed using MF on a pilot scale. Certain variability in the MF process affects operational costs. A pressure of 1.95 bar, 1.0 g DCW/L initial biomass concentration, 0.70 kWh, 25 °C, and a membrane area of 3.8 m² need operating costs of 2.86 kWh/kg of biomass. Changes in certain operating variability can reduce operational costs significantly to 1.27 kWh/kg of biomass with a pressure parameter of 1.95 bar, 2.0 g DCW/L initial biomass concentration, 0.46 kWh, 20°C and membrane area 7.6 m² (Gerardo et al., 2015).

The results of previous studies indicate that water treatment with MF/UF is more economical than conventional processes (Al-Malack, 2003). The MF crossflow process can produce reliable permeate quality. Meanwhile, in conventional processes, water quality depends largely on operational factors. The operational difficulties of the MF/UF process are also less than those of the conventional process. Capital costs from conventional water treatment processes are

65% greater than using MF/UF, while annual operating and maintenance costs are 21% lower. Based on predictions, after MF/UF is used for 40 years, it will be 8% more efficient than conventional processes.

Based on the study, the NF system has a capex (capital expense) estimated at US\$ 551,250.00 and an opex (operating expense) ranging from US\$ 0.364 to 0.446 per cubic meter of wastewater for an estimated membrane life of 1 - 5 years. Industry can save more than US\$ 99,000 per year by reusing permeates. It can reduce water consumption around 551,880 m³ per year (Reis et al., 2019). The bioreactor-NF system on a pilot scale records unit costs per m³ of wastewater treatment for US\$ 2.57, not including labor costs (Li et al. 2020). If implemented at full scale, the results will show a better efficiency of wastewater treatment, which has good quality for reuse (He et al., 2013). The NF process shows very good potential in waste treatment and waste reuse.

In other studies, NF/RO works well by rejecting an average of 94.5 – 99% of the metal and giving a residual concentration of 0.01 – 0.77 g/L for Cu and Ni and 0.7 – 5.7 µg/L for Zn. Operational costs consist of energy demand (47%), membrane replacement (37%) and for others (16%). NF/RO reliably removes metals below 6 µg/L with an operating cost of 18 p/m³ (Garcia et al., 2013). NF/RO technology is increasing globally because there are always continuous improvements to significant cost reductions. These advances include membrane properties, module design, process design, initial feeds, energy recovery devices, and operational strategies focused on reducing energy consumption (Peña et al., 2013; Rahimpour et al., 2010).

4. CONCLUSION

The MF/UF system can serve as an effective pre-treatment for the NF/RO process, particularly for removing bacteria. Moreover, MF/UF membranes can be modified to enhance their performance, improving filtration. In contrast, the NF/RO system can eliminate a broader range of contaminants than MF/UF, including ions, heavy metals, viruses, and pharmaceutical compounds (PhAC). However, NF/RO membranes are more susceptible to fouling, which can be mitigated through antifouling strategies, such as developing NF membranes with zwitterionic functionalized monomers. RO, in particular, is highly effective for seawater treatment, providing high-quality water suitable for consumption. Additionally, water treatment using NF and RO systems presents a promising alternative due to their ability to operate with lower energy consumption than conventional water treatment methods. However, cost considerations play a significant role, as the water treatment process using a single MF/UF system is more economical than a combined MF/UF and NF/RO process, though the latter cannot remove all types of contaminants as effectively as NF/RO. This review highlights the need for further research into low-cost antifouling membranes and the integration of renewable energy to support the sustainable implementation of NF/RO systems. Policymakers and engineers should consider these insights when designing scalable, context-appropriate water treatment solutions.

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